

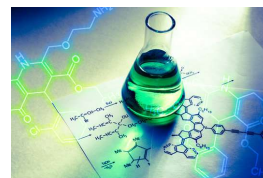
SCHOLARSHIP OF EXCELLENCE IN TEACHING FELLOWSHIP

Strategies For Students Engagement

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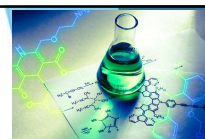


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Goal :

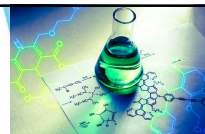
Enhance Students learning by creating an environment to engage with each other and learn from one other.

Deepen students understanding of the important role of chemistry in everyday life experiences and processes



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Why this goal is important



Low participation during class discussions

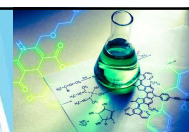
Importance of student- student learning

Importance of enhancing students' skills in the area of research, data collection & analysis, critical thinking, and collaboration

Improving students understanding of the role of chemistry in real-world systems and processes

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Scholarship that supports this strategy



Engagement is a multidimensional process

Behavioral engagement (participation, effort)

Emotional engagement (interest, enjoyment)

Cognitive engagement (deep learning, investment in understanding)

True engagement includes all three

Elizabeth Barkley, Student Engagement Techniques

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Scholarship that supports this strategy

Motivation is central to engagement

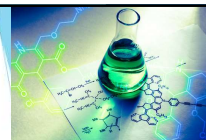
Students need both **value** (why this matters) and **expectancy** (belief they can succeed).

Instructors to make **relevance** visible and to **design** achievable but challenging learning tasks.

Collaborative learning enhances engagement

Many Student Engagement Techniques (SETs) use **teamwork**, because structured collaboration builds community, accountability, and deeper understanding.

Elizabeth Barkley, *Student Engagement Techniques*



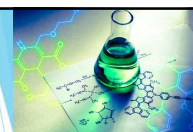
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Scholarship that supports this strategy

Teaching is inherently relational

Learning occurs within a complex web of relationships—**teacher-to-student**, **student-to-student**, and both connecting to the **subject matter**. Healthy relationships support deep learning.

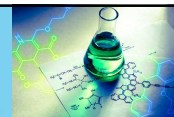
Parker Palmer, *The Courage to Teach*



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General Chemistry II (CHEM-132) Student Projects:

Project 1: Understanding and Conserving Household Water Consumption



Purpose:

The purpose of this project is to help students evaluate how much water is used in the household and to identify practical ways to conserve water consumption. The aim of this project is to promote environmental awareness and to encourage sustainable living habits.

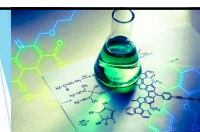
Learning Objectives:

- Measure daily and weekly water usage across various household activities.
- Analyze data to determine patterns and areas of high consumption.
- Research and propose effective water conservation methods.
- Develop a practical water-saving plan for the students' home.

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General Chemistry II (CHEM-132) Student Project-1

Instructions/Tasks



Data Collection: Record water usage for daily household activities like bathing, cooking, washing clothes, cleaning, etc. You can use tools like water bills, water flow meters, or manual estimates for measurement.

Data Analysis: Calculate total daily and weekly water consumption of each household activity. Identify which activities consume the most water

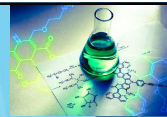
Research: Study water conservation methods such as behavior changes, leak detection, low flow fixtures, etc. Study environmental and economic impacts of excessive water consumption.

Action plan: Suggest 3-5 water saving strategies. Predict potential water savings if these strategies are implemented.

Presentation- Report: Prepare a summary report of your findings, and a slide presentation. Prepare a brochure that provides a simple water-saving guide to share with fellow students and community

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General Chemistry II (CHEM-132) Student Projects:



Project 2 : Wastewater treatment Process

Purpose:

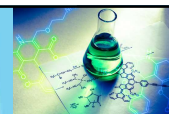
The purpose of this project is to help students gain an understanding of wastewater treatment process and operational parameters the mean cell residence time (MCRT). In wastewater treatment, MCRT is a useful parameter that represents the average time microorganisms that break down waste spend within the activated sludge process.

Learning Objectives:

- Gain an understanding of the numbers and units involved in wastewater treatment calculations.
- Use representative data and calculate volumes, times, and chemical demand
- Develop a better connection to our wastewater treatment facilities, and learn how they operate on a basic level.

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General Chemistry II (CHEM-132) Student Project-2



Instructions/Tasks

Overview of Wastewater Treatment: A brief lecture/description of operations at a wastewater treatment facility. Discuss the basic processes and their respective timelines.

Data Review: Given a set of quantities relevant to wastewater, study and understand them. These data will be given in the context of the mean cell residence time.

Data Analysis: Calculate the mean cell residence time using the given data.

Research: Look at the MCRTs for a variety of facilities in the US and compare based on volume and demand; much of this can be found on local water system websites.

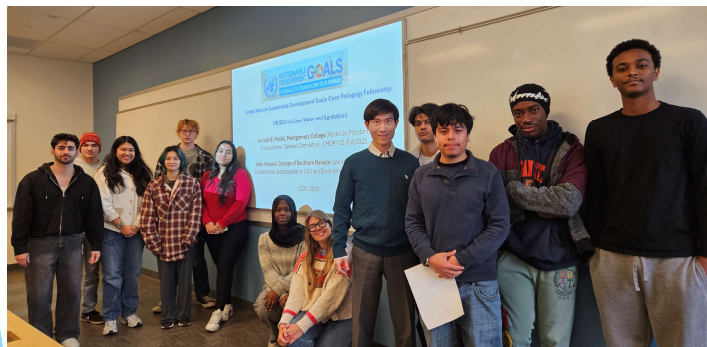
Action plan: Suggest how one could increase/decrease MCRT, discuss the pros/cons and challenges one would face in attempting to manipulate the MCRT at a given facility.

Presentation- Report: Prepare a slide presentation that discusses MCRT, with 3 examples from your research. Include elements from the action plan highlighting the challenges of this particular aspect of wastewater treatment.

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General Chemistry II (CHEM-132- Fall 2025) Student Projects:

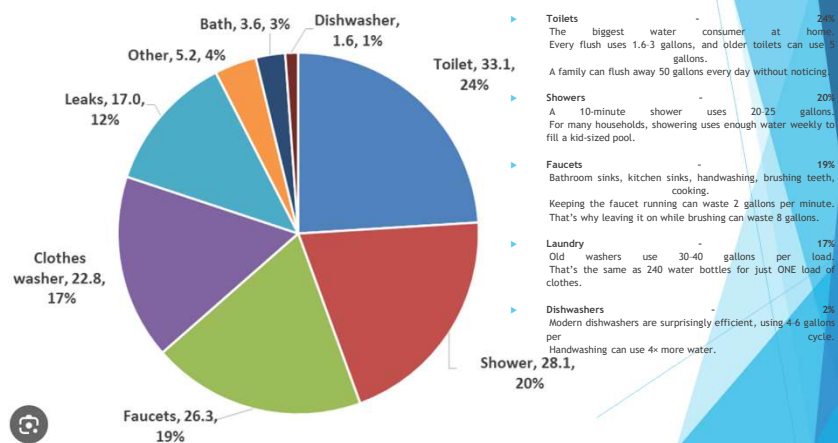
Choi	Zechariah	Mesgari Khosrosha	Helya
Delgado	Xavier	Ortiz Canales	Ashley
Feleke	Aaron	Padilla Vidal	Jennifer Carolina
Fishell	Charis	Perez	Alexander
Graham	Derek	Perez	Moises
Jack	Fatoumatta	Peter	Anurag
Lasko	Parker	Psaltis	Roxanne
Lessard	Allyson	Rai	Adishree
Maharjan	Erica	Shaposhnikov	Matvei
Melton	Samantha		



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General Chemistry II (CHEM-132) Student Project-1

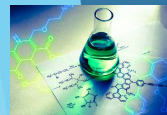
American household water usage



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General Chemistry II (CHEM-132) Student Project-1

Helya's data table



Activity / Fixture	Typical Usage / Flow Rate	Activity	Week 1 min	Week 1 gal	Week 2 min	Week 2 gal	Week 3 min	Week 3 gal	Week 4 min	Week 4 gal	% Change (Min)	% Change (Gal)
Washing machine (per load)	24-35 gallons per load	Laundry	2 loads	59.0	1 load	29.5	2 loads	59.0	1 load	29.5	—	—
Shower	2.5 GPM (low-flow)	Baths/Showers	65	162.5	80	200.0	45	112.5	35	87.5	—	—
Bathroom sink	2.0 GPM	Bathroom faucet	15	30.0	0	0.0	15	30.0	20	40.0	—	—
Kitchen sink / faucet	2.0 GPM	Dishwasher	60	5.68	65	6.15	45	1.89	15	1.42	—	—
Dishwasher (normal cycle)	5.3 gallons per cycle	Kitchen faucet	45	90.0	40	80.0	35	70.0	25	50.0	—	—
Water filter (tap-mounted)	0.4 GPM	Water filter	19	7.6	24	9.6	12	4.8	7	2.8	—	—
Refrigerator ice & drinking water	1.19 gallons/day per household	Refrigerator	7 days	8.33	7 days	8.33	7 days	8.33	7 days	8.33	—	—
Toilet flush	1.6-3.5 gallons per flush	Outdoor / Carwash	0	0.0	20	160.0	0	0.0	0	0.0	—	—
Outdoor hose / carwash	8-10 GPM (garden hose or commercial carwash)	Plants watering	1.25	1.25	1.25	1.25	1.25	1.25	1.25	1.25	—	—
Plants watering	-2 cups per plant per watering (0.125 gallons/plant)	Total / Average	204	364.36	229	494.83	127	287.77	102	220.8	+12.25%	+35.73%
Boiling water / cooking	-1.2 gallons per day per person (depends on cooking habits)	% Change vs Week 1	—	—	+12.25%	+35.73%	-37.75%	-21.04%	-50.00%	-39.41%	—	—
Hand washing	0.5-1 gallon per hand wash (= 3-5 min at 2 GPM)											

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General Chemistry II (CHEM-132) Student Project-1

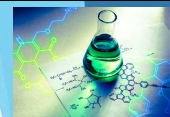
Recommendation for the Public

- ▶ Purchase and implement water conserving utilities such as low water use appliances like dishwashers, toilets, and more.
- ▶ Utilize small water conservation habits such as showering for less than 20 minutes a day or not running sinks and faucets high when washing dishes.
- ▶ Research what water systems your neighborhood utilizes to determine what larger water conservation utilities and appliances you could implement.



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General Chemistry II (CHEM-132) Student Project-2



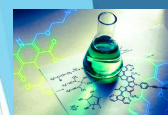
Comparison of Key Process Parameter

(approximate values based on MGD)

City	MLSS (mg/L)	Volume (MG)	WAS Conc (mg/L)	WAS Flow (MGD)	Plant Flow (MGD)	MCRT (days)
Las Vegas	2000	0.5	5000	0.05	3.0	3.58
Washington DC	3200	2.5	9000	0.1	5.0	8.43
Phoenix	2600	1.2	7000	0.1	4.0	4.22
New York	3000	3.0	8500	0.18	7.0	5.63

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Selected Feedback from students



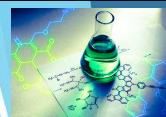
"I liked how the project applied general chemistry concepts to a real-world system that directly affects us"

"Using realistic values made the calculations more meaningful and helped me see how unit conversions and chemical dosing are used outside the classroom"

"The project was beneficial because it's strengthened my understanding of concentration and multi-step calculations while showing how small chemical amounts can have a large impact at the community level."

"Working on this project increased my engagement with the course by connecting chemistry to real life applications"

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As a result of SET program and implementation of this strategy

- 1) Students worked together to successfully complete examples of Real World and Relevant projects
- 2) Students enhanced their understanding of Household Water Consumption & Conservation, and Municipal Water Treatment Process and how chemistry plays an important role in such processes
- 3) Students practiced and improved their skills in the following area:
 - Literature search and research methodology
 - Data collection and data analysis
 - Effective communication and presentation skills
 - Collaboration and Teamwork

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Thank You !



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